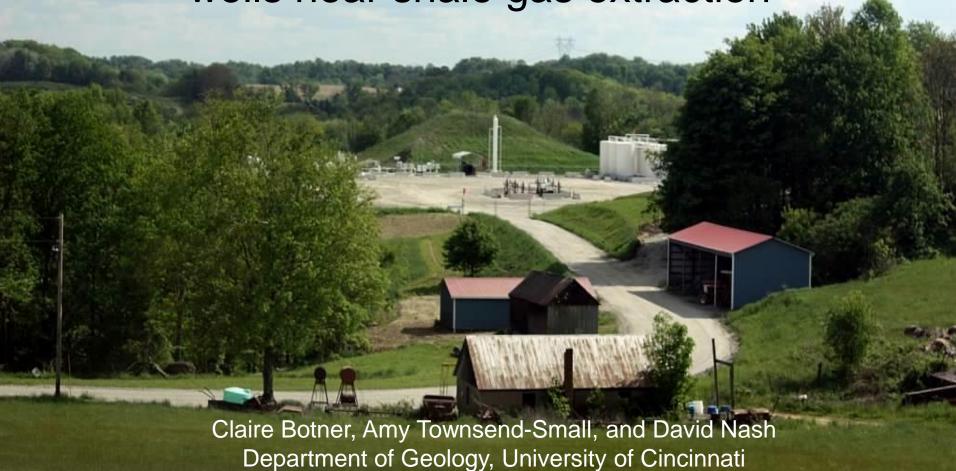
Flevated methane levels from biogenic coalbed gas in Ohio drinking water wells near shale gas extraction



Paul Feezel
Carroll Concerned Citizens

Natural Gas

- Recent glut in natural gas production in U.S.
- Is natural gas a bridge between traditional fossil fuel sources and renewable energy?
- Primary component is methane (CH₄) – a greenhouse gas 34x more powerful than CO₂

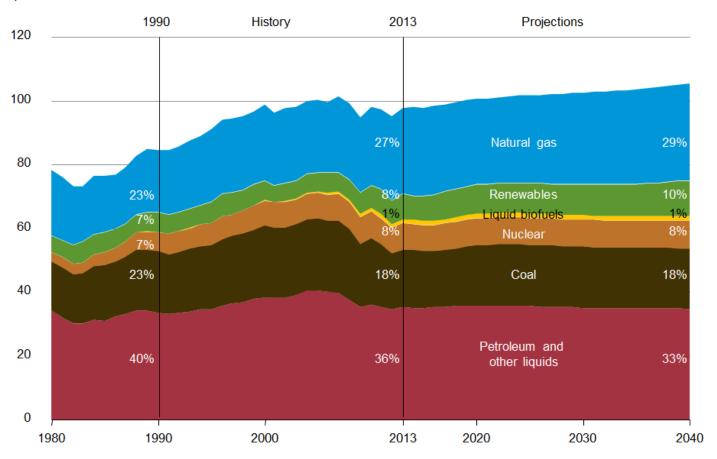


warriorpublications.wordpress.com

Coal/Oil-To-Gas Transition

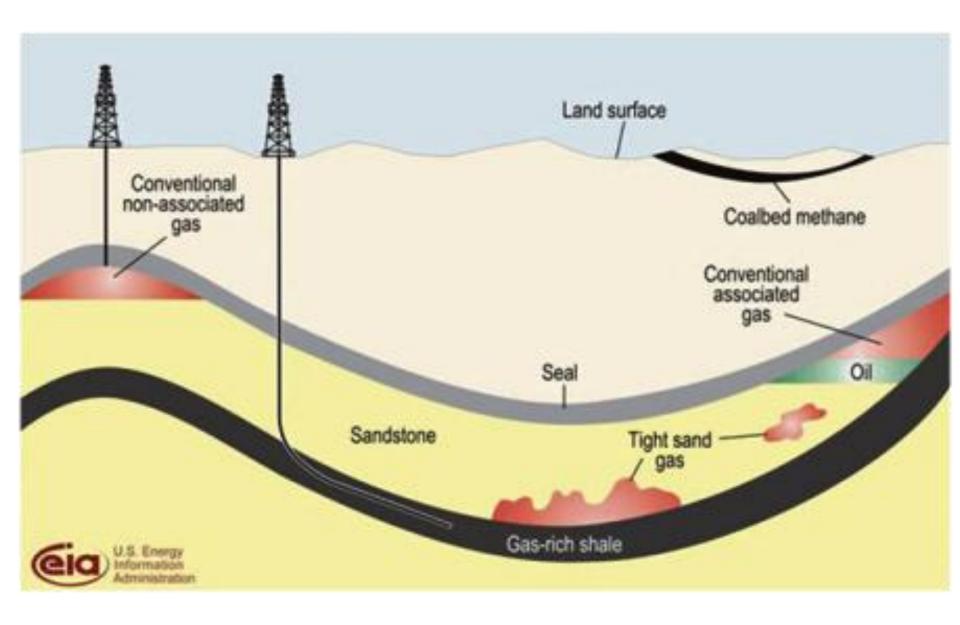
Figure 18. Primary energy consumption by fuel in the Reference case, 1980-2040

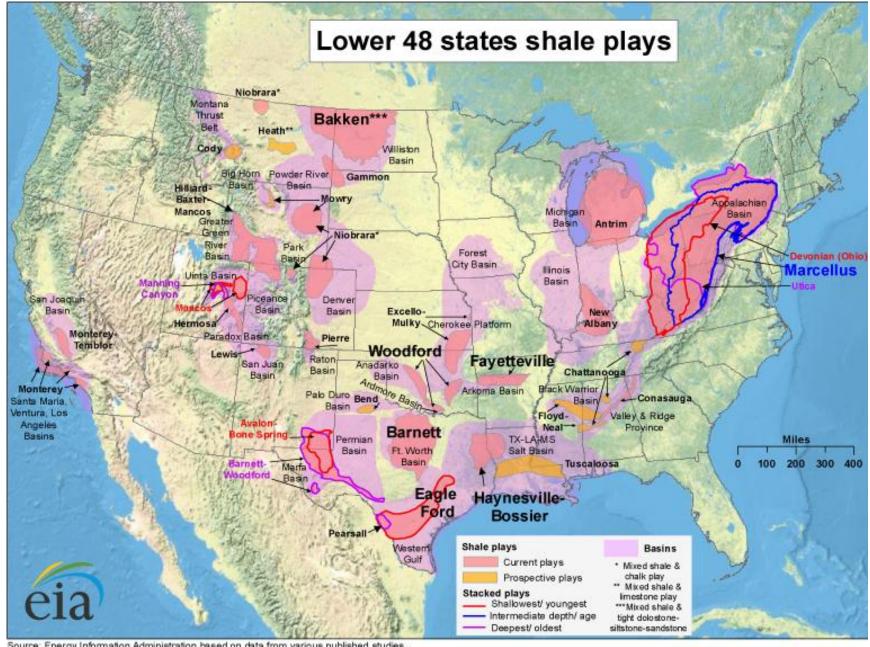
quadrillion Btu





Conventional vs. Unconventional Drilling





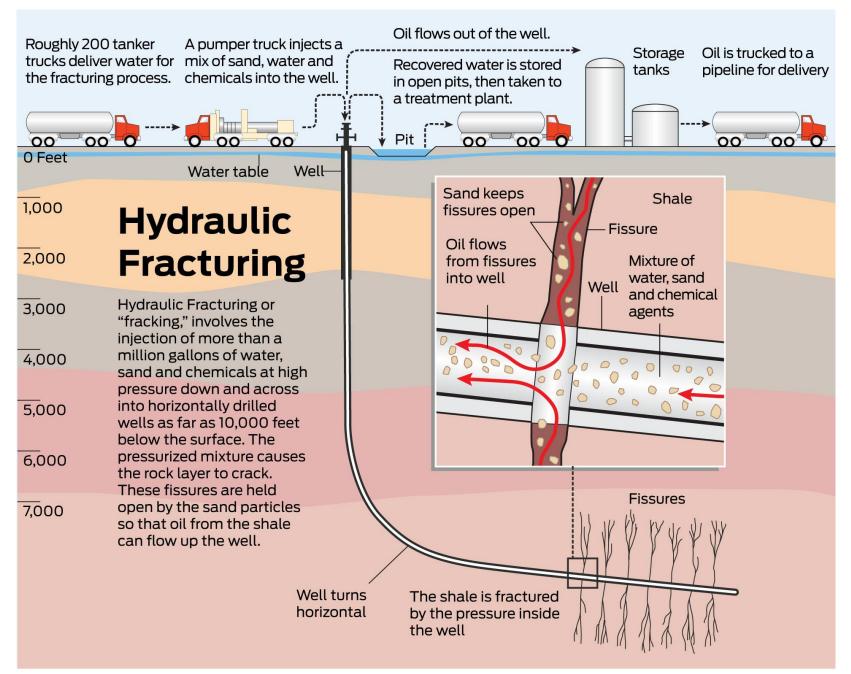
Source: Energy Information Administration based on data from various published studies. Updated: May 9, 2011

How Fracking Works

- Vertical drilling ~3 km + horizontal drilling up to 1 km
- 1. Detonation of explosives in well after drilling
- A mix of water, sand, and chemicals are pumped at very high pressure into the well to release gas from fissures



Photo credit: Casey White



Potential Pathways of NG Contamination

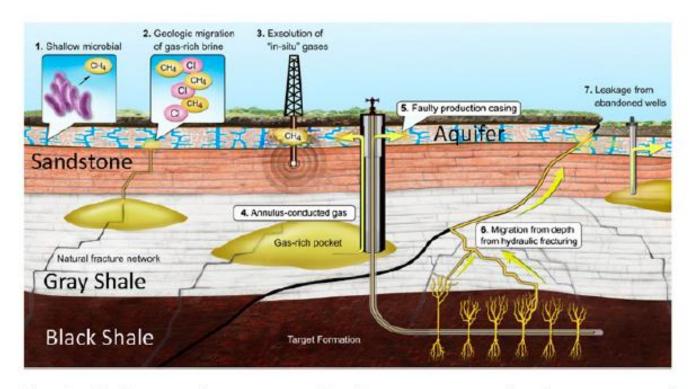


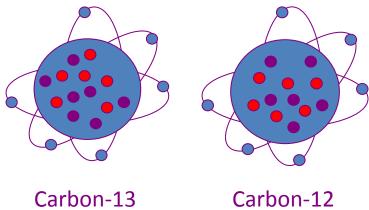
Fig. 1. A diagram of seven scenarios that may account for the presence of elevated hydrocarbon gas levels in shallow aquifers (see discussion in text). The figure is a conceptualized stratigraphic section and is not drawn to scale. Additional scenarios (e.g., coal bed methane and natural-gas pipelines leaking into aquifers) are unlikely in our specific study areas (Figs. S2 and S3).

Methane: Where can you find it?

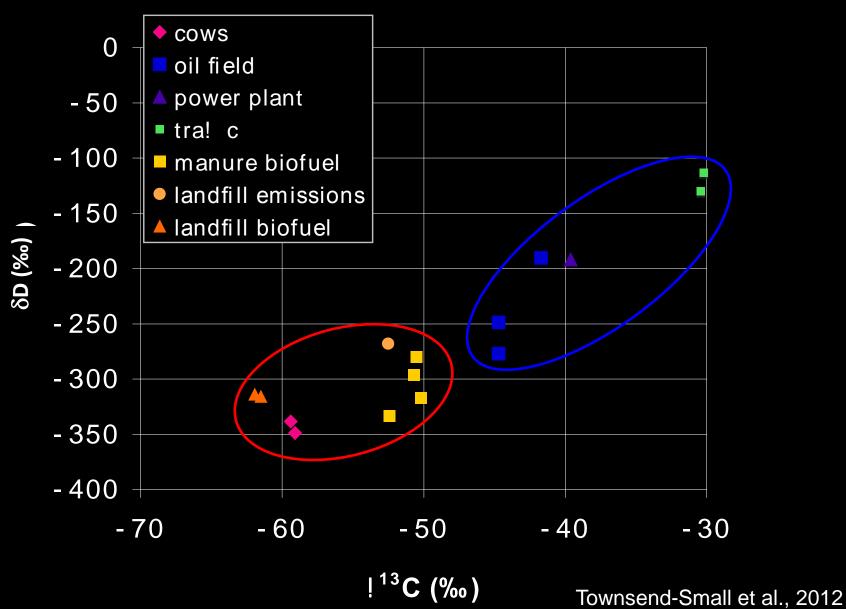
- Natural gas: ~95% CH₄
- Two main forms of CH₄
 - Thermogenic (created from intense heat and pressure of organic matter deep in the Earth; i.e. natural gas, oil)
 - Biological (created from microbial methanogensis; i.e. landfills, ruminant animals, wetlands, rice paddy agriculture)
 - Acetate fermentation
 - Carbonate Reduction

Stable Isotope Primer

- Isotope: Atom of a specific element that differs in the number of neutrons in the nucleus
- Isotopes with an extra neutron have the same chemistry but are slightly heavier than other atoms of the same element
- Stable isotopes of methane:
- Hydrogen
 - Hydrogen-1 (99.98% of all H atoms)
 - Hydrogen-2 (deuterium or D) (0.01%)
- Carbon
 - Carbon-12 (98.9%)
 - Carbon-13 (1.1%)

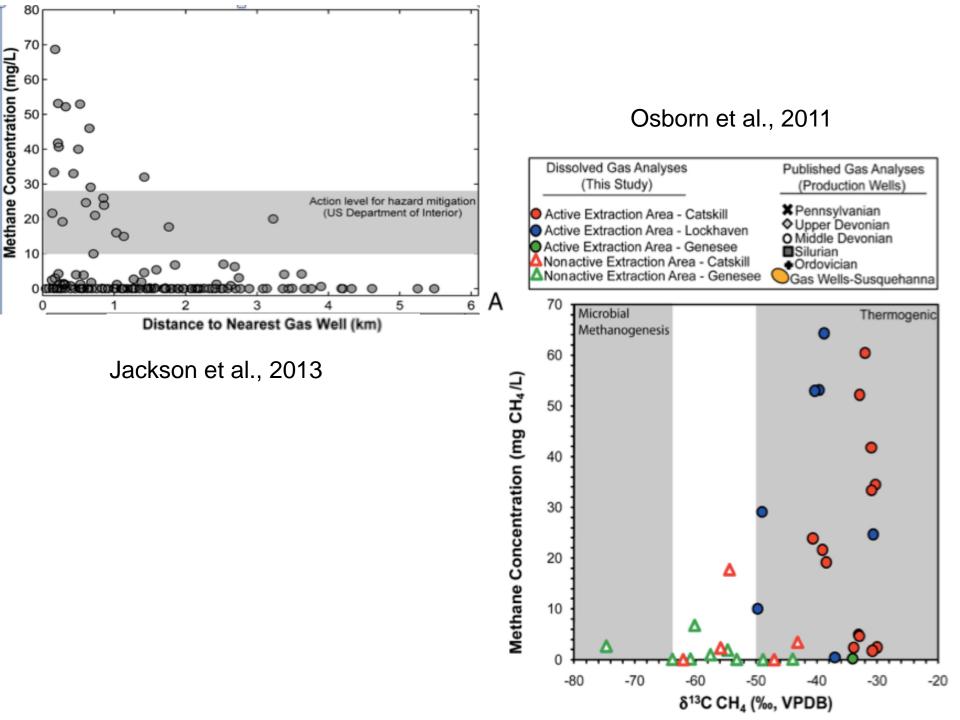


C and H stable isotopes of $\overline{\text{CH}_4}$



What is the problem?

- Is hydraulic fracturing contributing to methane found in groundwater resources of the Utica Shale?
- Previous studies (Osborn et al., 2011; Jackson et al., 2013) found that natural gas was present in groundwater resources after the onset of fracking in the Marcellus Shale
 - Utilization of stable carbon isotope analysis
- Our study aims to measure methane and its sources in groundwater before, during, and after the onset of fracking



Fracking and water resources

- Each well takes several million gallons of water to "frack"
 - Additives
 - Sand
 - Acids
 - Biocides
 - Salts
 - Antifreeze
 - Corrosion inhibitors
 - Some specific chemicals are proprietary and drillers need not disclose their formulas, although many do voluntarily



Fracking and water resources

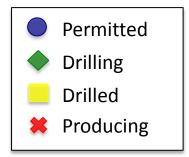
- After fracturing, water is returned to the surface with additional solutes from interactions with the shales
 - Salts
 - Methane
 - Hydrocarbons (benzene, etc.)
 - Radioactive materials
- Disposal or reuse

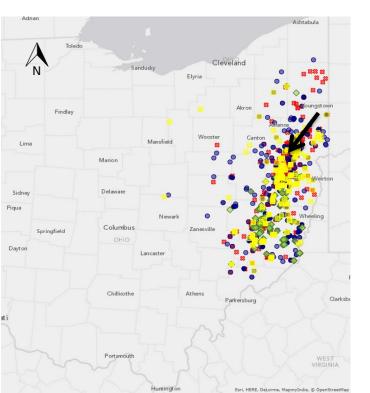


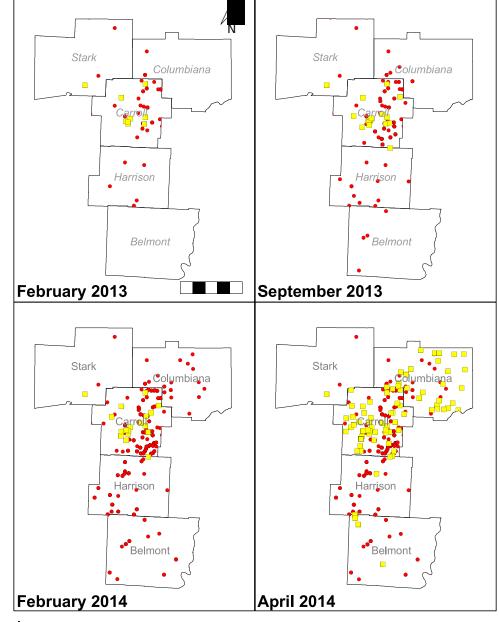
youtube.com

Highly viewed YouTube video showing drinking water with high levels of methane

Study Area







Legend

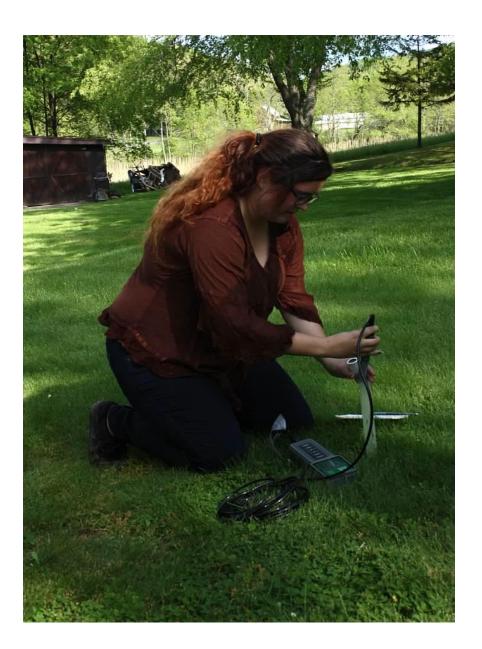
- Water Well
- Active Gas Well

Methods

- Teamed with Carroll Concerned Citizens to recruit participants
- Private groundwater well sampling in Carroll County, Ohio and surrounding area
 - 23 wells sampled from 2012 to Feb. 2015 three to four times a year
 - 191 samples total
 - Larger campaign in May 2014



What was measured?



- CH₄ Concentration
- Stable isotope composition of methane
 - $-\delta^{13}$ C and δ D
- pH and Conductivity
 - Indicators of fracking fluid contamination
- Δ¹4C analysis



University of Cincinnati's Stable Isotope Facility

Carbon and hydrogen isotope analysis performed with an isotope ratio mass spectrometer



Dr. Amy Townsend-Small

Department of Geology University of Cincinnati PO Box 210013

Cincinnati, OH 45221-0013

500 Geology/Physics Bldg. Phone: (513) 556-3763 Fax: (513) 556-6931

Date: May 22, 2014

Participant names: John Doe

Address: 100 Main Street Carrollton, OH 44615

Water Sampling Test Results

Introduction: The following data were gathered during a research project investigating impact of hydraulic fracturing ("fracking") on groundwater resources in Carroll County, Ohio. These results will not be associated with your name, address or well location on any scientific publications, nor will your personal information be shared with anyone else. Questions? Contact Dr. Townsend-Small.

Sampling Date	Туре	Electrical conductivity (microslemens per centimeter)	рн	Temp.	Total dissolved organic carbon (milligrams per liter)	Total dissolved nitrogen (miligrams per liter)	Dissolved methane (milligrams per liter)	Carbon Isotopic composition of methane (5 ¹³ C; ‰)	Benzene, toluene, ethylbenzene, and xylene (BTEX) (micro- grams per liter)
3/8/2014	G/W	753	NM	12.8	0.7	0.3	4.123	-66.5	NM
5/25/2014	G/W	863	8.24	16.9	0.4	0.3	18.610	NM	NM

NM = not measured

Explanation of Results

Type: GW = groundwater, SW = surface water (probably collected at a stream near your house)

<u>Electrical conductivity</u>. This is a measurement of the total amount of dissolved salts in your water. Most groundwater in this area has conductivity between 100 and 500 mS/cm. Measurements in the 1000 to 10,000 uS/cm range may indicate salty brine water is present.

pH; The acidity of your water. A pH value of 7 indicates neutral acidity. Most wells in this area have a pH of between 6 and 8.5. A lower pH may indicate the presence of additional acid.

<u>Total dissolved organic carbon</u>: This indicates the abundance of organic carbon molecules in water. Most unpolluted waters have values between 0.5 and 10 mg/L.

<u>Total dissolved nitrogen</u>: The total amount of dissolved nitrogen in your water. This could be derived from natural materials in soil, from fertilizer, or from human or animal waste. Most unpolluted waters have values between 0.05 and 0.7. Higher values may indicate the presence of fertilizer or wastewater.

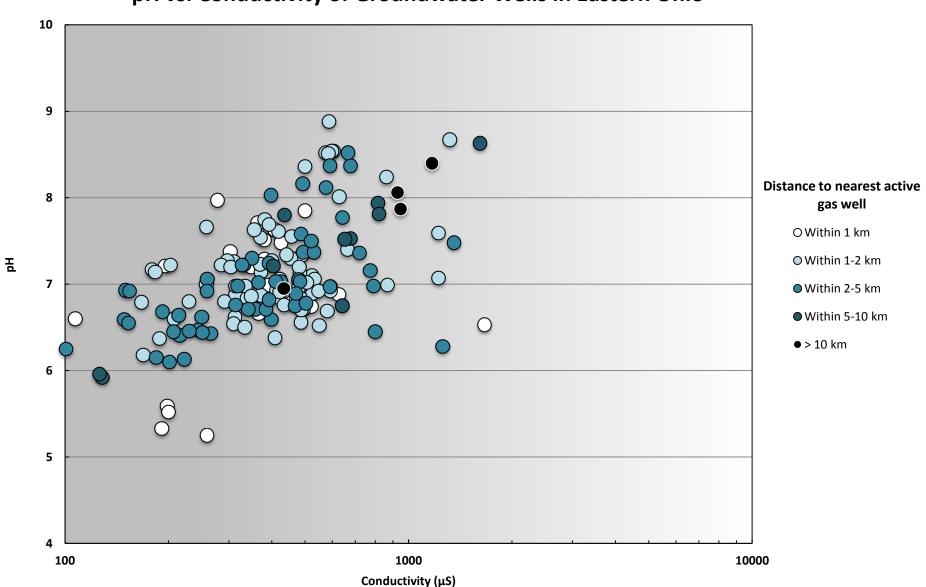
<u>Dissolved methane</u>: Methane is the primary component of natural gas, and is also naturally produced by harmless microbes living in oxygen-free environments, such as at the bottom of lakes or in mud. The presence of methane in groundwater is normal. Methane levels over 10 mg/L may indicate the presence of natural gas, but this can be confirmed with stable isotope analysis.

Carbon isotopic composition of methane: This is a way to determine the source of methane. Methane derived from natural gas has a value between -45 and -30%. Methane from naturally occurring microbes has a value between -70 and -50%. We are currently analyzing a backlog of samples for carbon isotopes so more data are forthcoming.

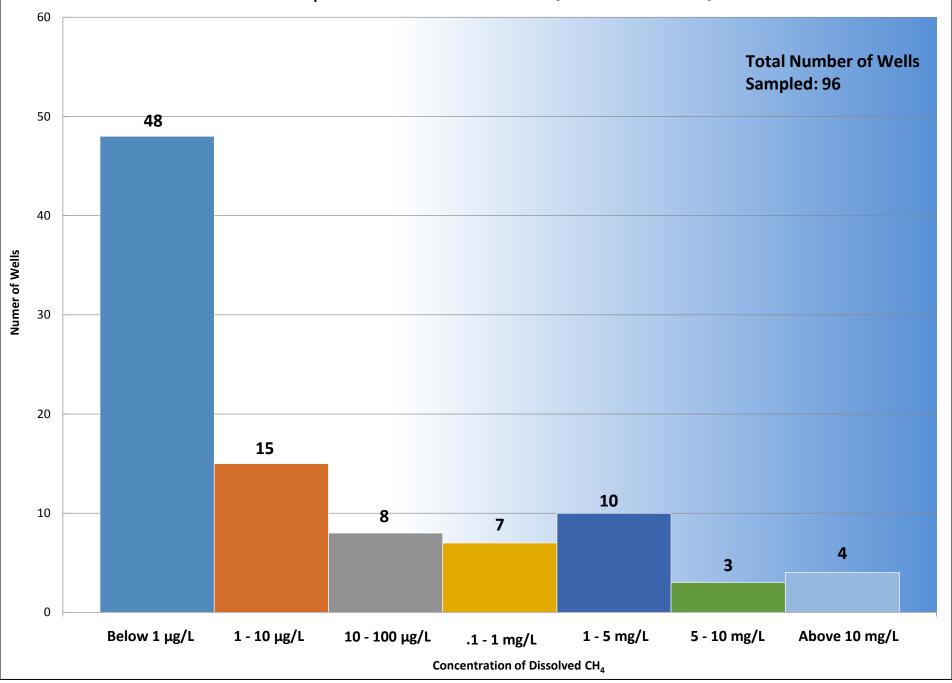
Benzene, toluene, ethylbenzene, and xylene: This suite of chemicals (sometimes referred to as BTEX) represents the carcinogenic component of natural gas. Any BTEX chemical present in your water represents a health hazard. We are currently seeking more funding for further analyses of these chemicals.

Results

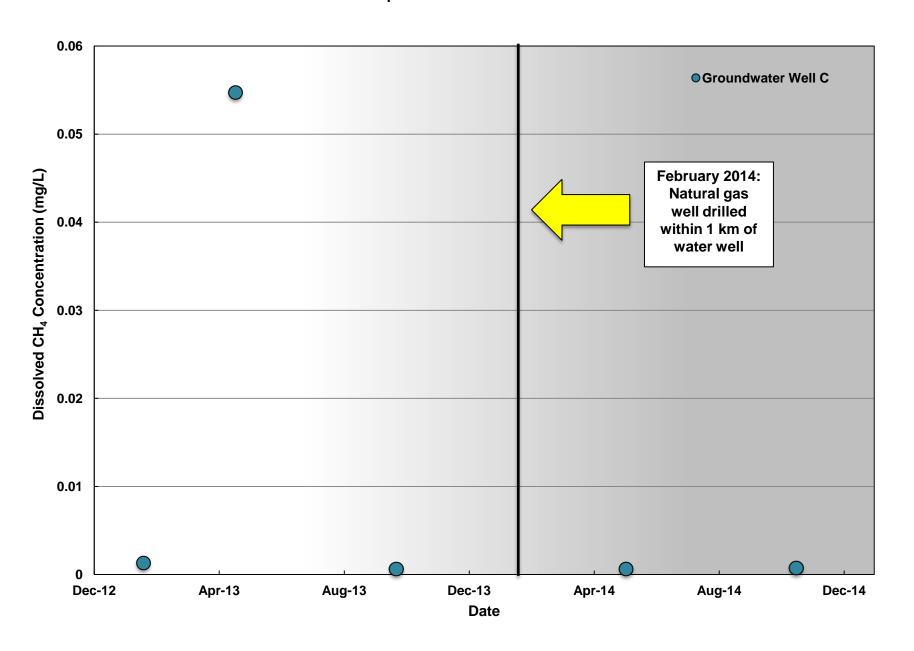
pH vs. Conductivity of Groundwater Wells in Eastern Ohio



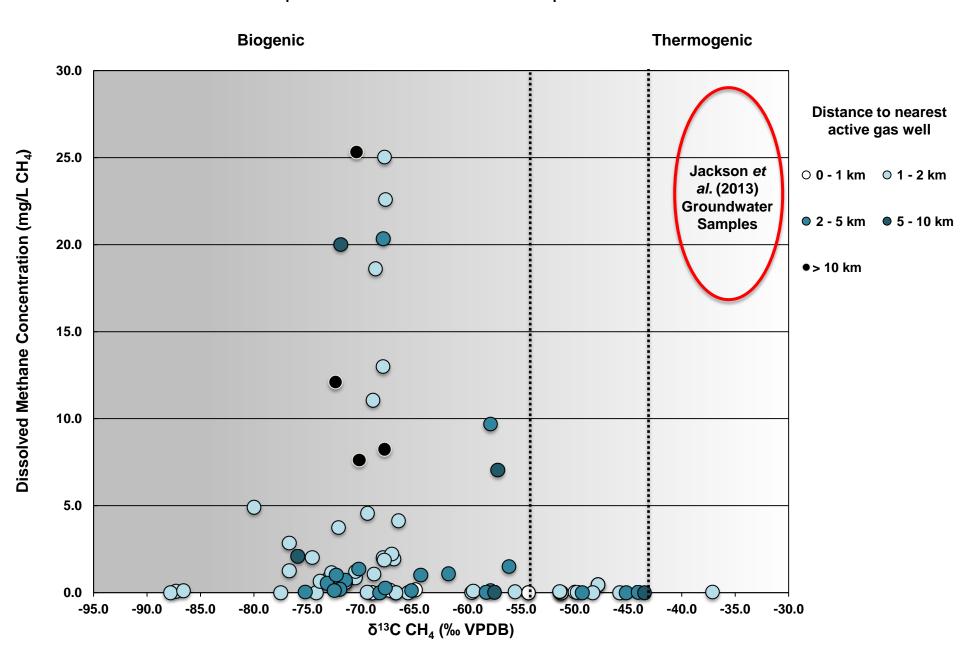
Dissolved CH₄ Concentration in Participant Wells, May 2014



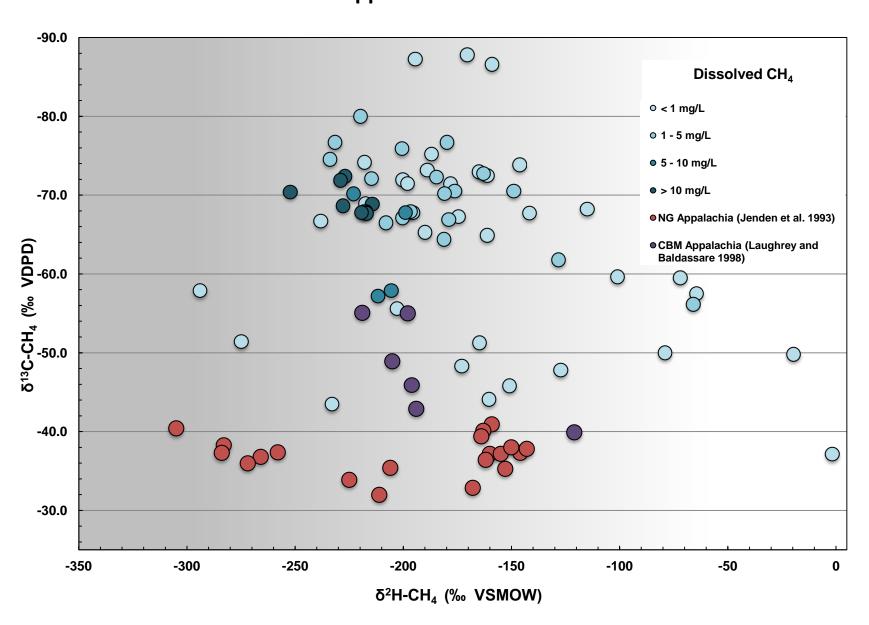
Time Series of CH₄ Levels in Groundwater Well "C"



Dissolved CH_4 Concentration vs. $\delta^{13}C$ CH_4 of Groundwater Wells



 $\delta^{13}\text{C-CH}_4$ vs. $\delta^2\text{H-CH}_4$ of Groundwater Samples and Previously Reported Appalachian Natural Gases



Radiocarbon Primer

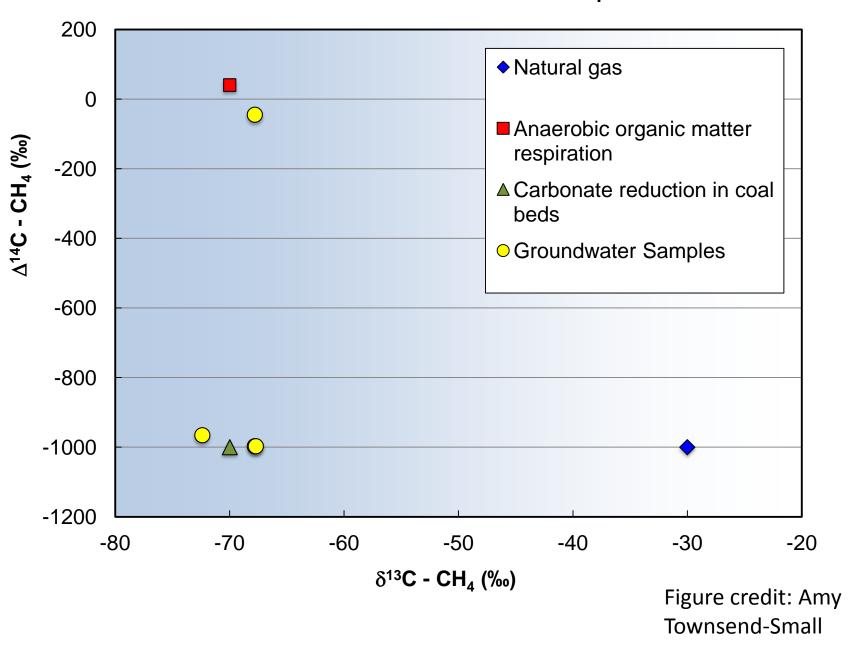
• Sun \rightarrow ¹⁴N to ¹⁴C \rightarrow ¹⁴CO₂ \rightarrow Plants \rightarrow Animals

- "Modern" carbon: contains measureable ¹⁴C
 - Positive Δ¹⁴C values
 - Methane originating from bacteria

- "Dead" carbon: undetectable level
 - Reaching close to -1000 ‰
 - Coalbed CH₄



Δ^{14} C vs. δ^{13} C of various CH₄ sources



Conclusions

- Some water wells contain very high levels of CH₄
 - Explosive range ~10 24 mg/L CH₄
- δ¹³C values indicate that CH₄ found in groundwater has biological origins
- Groundwater contamination from fracking has happened, but it isn't the norm
- Normal pH and conductivity values show that fracking fluid is likely not present in sampled groundwater resources

Conclusions & Future Work

 Δ¹⁴C analysis of four select wells indicates high levels of CH₄ can be attributed to biogenic coalbed gases in at least three of the wells

Further characterization of coalbed gas in eastern Ohio groundwater

Long-term monitoring necessary

Acknowledgements

- Dr. Amy Townsend-Small
- Dr. David Nash
- Paul Feezel, Carroll Concerned Citizens
- Kristine Jimenez
- Deer Creek Foundation
- Alice Weston Foundation





Questions?



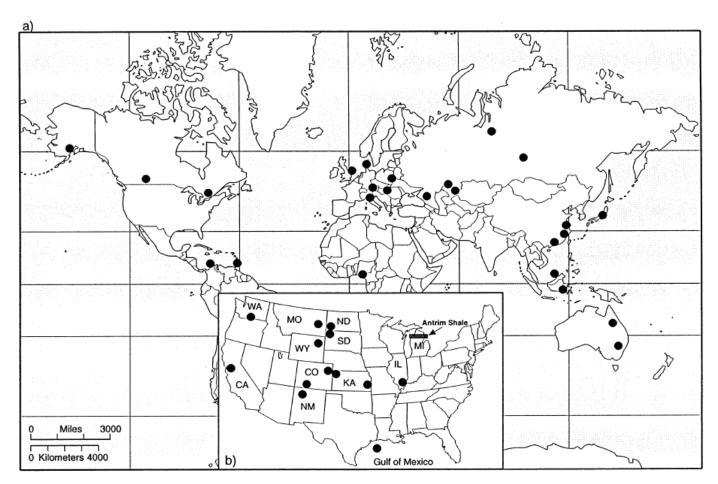


Fig. 1. Methane occurrences of known bacterial origin: (a) Worldwide and (b) United States (modified from Rice, 1993). Location, reservoir age, C + H isotope compositions and original source of data shown as solid circles are in Rice (1993). More recent biogenic methane occurrences are added; for coalbed methane from the San Juan Basin, New Mexico, U.S. (Scott et al., 1993), the Bowen and Sydney basins, Australia (Smith and Pallasser, 1996), and for the Antrim Shale (grey rectangle), Michigan, USA (Martini et al., 1996).

Martini et al. 1998